Production of biodegradable starch nanocomposites using cellulose nanocrystals extracted from coconut fibers

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Abstract

Different polymeric matrices have been investigated for use in the development of biodegradable films. The incorporation of cellulose nanocrystals in such films has particularly attracted attention because of the potential for achieving improved properties of starch nanocomposites. In the present study, cellulose nanocrystals were extracted from coconut fibers and incorporated in cassava and potato starch films at different concentrations. The properties of the different nanobiocomposite films were comparatively evaluated, including their barrier and mechanical properties. All the films, regardless of the nanocrystal concentration, were found to exhibit low solubility in water, with increased moisture content particularly observed in the films with higher nanocrystal concentrations. The potato starch film with the lowest nanocrystal concentration was found to exhibit the best mechanical properties. The observations of this study indicated that the source of the starch and the nanocrystal concentration determined the properties of the nanobiocomposite films.

Keywords: lignocellulosic fiber, biodegradable film, polymeric matrix, nanocrystal.

1. Introduction

Plastics are the most commonly used materials in the food industry, owing to their moldability, manageability, low cost, and suitable mechanical and chemical properties^[1]. However, the diversity of resins used for their production poses challenges to their separation for reuse^[2]. This has contributed to the environmental problems associated with discarded plastic packaging, with the significant buildup of residues requiring about 100-450 years for their natural decomposition^[3]. The increasing demand for environmental preservation has thus necessitated research into alternatives to conventional plastics, such as varieties derived from petroleum derivatives using biodegradable and ecologically friendly materials.

The technological adaption of biomaterials, particularly polymers, for the development of flexible films has been widely studied in recent years^[4–6]. Several biopolymers such as polysaccharides, proteins, and lipids have been utilized as polymeric matrixes for the development of alternative biodegradable packaging, owing to their availability, renewability, low cost, environmental friendliness, and biodegradability^[7-9]. Among such agriculturally sourced materials, starch is considered to be favorable for the development of biocomposites^[10,11]. Starch films are particularly characterized by transparency, non-toxicity, and low cost. However, they also have some limitations, the most prominent of which are low flexibility, high permeability to water vapor^[2,12], and inadequate mechanical and barrier resistance. This causes them to fall short of market expectations, especially in the food industry. Consequently, several studies have been conducted on the use of reinforcing additives such as fibers^[13,14] and cellulose nanocrystals^[15-17] to improve the mechanical, thermal, and barrier properties of starch-based films.

Cellulose is the most important polymer and is used in various fields, especially because of its availability, biocompatibility, and biodegradability^[18,19]. Cellulose fibers have an inherent structural hierarchy originating from their differing biological sources. The production of cellulose nanocrystals from different lignocellulosic fibers has attracted interest in the scientific and industrial community in recent years. This has been owing to their mechanical and thermal properties, cost effectiveness, and versatility^[20,21]. Nanocrystals are generally characterized by high solidity and crystallinity, which afford sustainability. When incorporated in films with renewable polymeric matrixes, they produce total biodegradable composites^[22] referred to as nanobiocomposites. As with any composite material, the properties of the nanobiocomposites depend on the properties of the individual components (matrix and reinforcement), the composition (volumetric fractions of the components), the morphology of each constituent phase (spatial arrangement, dimensions, and crystallinity), and the interfacial properties^[23].

It is important to highlight that the development of new composites with natural fiber reinforcement is quite promising, bearing in mind that, in addition to the renewability of the fiber sources, they also cost less than synthetic fibers^[24,25]. Incidentally, Brazil is acknowledged as one of the largest agricultural producers in the world, contributing a large part of the millions of tons of lignocellulosic biomass produced globally from different sources^[23]. The green coconut bark is an important source of natural fiber, and its utilization for the production of cellulose nanocrystals offers the benefit of reduced disposal of biomass^[25-27].

Further, it should be noted that the use of biodegradable packaging in the food industry promises to enhance the preservation of the packaged products. It would also lead to the creation of new consumer markets as the differentiated biodegradable properties of the films fit new consumer profiles, especially those who are prepared to pay more for products or services with a green stamp (a certificate granted to products manufactured for sustainable development)^[28,29].

It is noteworthy that recent studies have highlighted the benefits of incorporating cellulose nanocrystals in flexible films produced from cassava starch, namely, the improved properties afforded by the produced biodegradable starch nanocomposites^[5,6,10,12]. In the present study, we incorporated cellulose nanocrystals produced from green coconut fiber in flexible films with polymeric matrixes of tapioca starch and potato starch, respectively. We evaluated the effects of the starch source and the concentration of the cellulose nanocrystals on the properties of the films, and determined the best matrix and nanocrystal concentration. Glycerol was used as plasticizer in the development of the flexible films. The objective of the study was to characterize green coconut fiber cellulose nanocrystals with respect to their size.

2. Materials and Methods

2.1 Extraction of cellulose from green coconut fiber

The extraction of the cellulose pulp from the green coconut fiber was performed based on the works of Rosa et al.^[26], Samir et al. [30], and Machado et al. [5]. The coconut fibers were dried in a greenhouse and ground in a liquidizer to obtain a fine particulate. The particulate was further dried and 30 g of it was washed by constant agitation for 4 h in 1.2 L of a solution of 2% NaOH at 80°C. The fine particulate was then filtered and the washing process was repeated three more times to completely remove the water-soluble agents. The filtered particulate was subsequently further washed in distilled water to obtain the cellulose pulp. This was followed by delignification (or whitening) of the pulp using a mixture of 0.3 L of 1.7% sodium hypochlorite and 0.3 L of a buffer solution. The cellulose pulp was constantly agitated in the mixture for 6 h at 80°C, and then filtered and dried in a greenhouse. It was finally pulverized in a mill (Cadense Ltd, Brazil).

2.2 Preparation of cellulose nanocrystals

The cellulose nanocrystals were prepared by acid hydrolysis using 64% H₂SO₄^[5,6,26], with 12 mL of cellulose pulp per gram of the acid subjected to constant agitation for 5 min at 50°C. After acid hydrolysis, the samples were filtered and their concentrations adjusted. The samples were then centrifuged for 10 min at 4400 rpm and 10°C to separate the crystals. This procedure was repeated eight times, when no more supernatant was produced. The suspensions were then dialyzed using the D9777-100 FTO cellulose membrane with 12.000 Da cut off (Sigma-Aldrich) until their pH was approximately 7, after which the samples were placed in an ultrasonic bath for 20 min.

2.3 Preparation of nanobiocomposites

The nanobiocomposites were prepared by the casting method, which is used for the preparation and drying of films. The process involved the preparation of a film-forming solution in which the polymeric matrix (4%, g/100 g) manioc starch (Cargill Agrícola SA) or potato starch-and glycerol (Synth[®], 1.0%, g/100 g) were initially dissolved. The solution was prepared with distilled water. The mixture was then added to a suspension of the cellulose nanocrystals containing varying amounts of the nanocrystals (0.5%, 1.0%, and 1.5% (g/100 g)) (Table 1). The new mixtures were then heated to the gelatinization temperature of the starch (~70°C) under constant manual agitation. 45 g of the mixtures was then weighed in polystyrene Petri dishes and dehydrated in a greenhouse under airflow (35±2°C) for 18-20 h. Other films were produced without the addition of cellulose nanocrystals for use as controls. Before their characterization, the prepared nanobiocomposites were stored at 60% humidity and 23°C for 10 days in a desiccator containing a saturated solution of sodium chloride.

2.4 Characterization of coconut fibers and cellulose nanocrystals

The main components of the fibers obtained from green coconut bark (lignin, hemicellulose, and cellulose) were characterized using the ANKOM A200 Fiber Analyzer. The methodology proposed by Van Soest et al.^[31] and Goering and Van Soest^[32] was employed. The fibers and cellulose were also characterized using a scanning electron microscope (SEM) (model JEOL JSM-6390LV). The fibers and cellulose samples used for the SEM characterization were metalized with gold in a "sputter coater" (model SCD 50, Balzers) using argon plasma, and then left to rest for 24 h.

To determine the birefringence of the suspensions of the cellulose nanocrystals in water, two films of crossed polarizers were used. The polarizer films were positioned perpendicularly to each other and a light was directed at one of them. The suspension sample was then interpolated between the two films^[33].

	Table 1.	Design	of the	experimental	nanobiocom	posites
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Formulation	Manioc starch	Potato starch	Cellulose nanocrystals	Glycerol (%, g/100 g)
EM05	(70, g100 g)	(70,g/100 g)	(70, g/100 g)	1.0
FIM05	4.0	-	0.5	1.0
FM10	4.0	-	1.0	1.0
FM15	4.0	-	1.5	1.0
FB05	-	4.0	0.5	1.0
FB10	-	4.0	1.0	1.0
FB15	-	4.0	1.5	1.0
Control FM	4.0	-	-	1.0
Control FB	-	4.0	-	1.0

The suspension of the coconut cellulose nanocrystals was examined by an SEM to determine the crystal length (*L*), diameter (*D*), and aspect ratio (*L/D*), and the state of the aggregation of the nanocrystals. The nanocrystals were visualized by coloring with 2% uranyl acetate (UA). The coloring was done by mixing the suspension of the cellulose nanocrystals with an equal volume of UA. 10 μ L of the mixture was then distributed through 400-mesh copper grids and left to stand for 30-60 s. The grids were subsequently dried and visualized by a transmission electron microscope (TEM) (BX-51, Olympus) operated in the clear field mode at 80 kV. The length and width of the nanocrystals were directly measured from the transmission electron micrograph. The mean and standard deviation for 30 measurements were determined.

2.5 Characterization of nanobiocomposites and controls

The thicknesses of each pre-packaged nanobiocomposite film (60% humidity, 25°C) were determined by calculating the average thickness for six measurements at random points. A flat-head digital micrometer (Mitutoyo) with a resolution of 1 μ m was used for the measurements. The water activity (aw) of the nanobiocomposite film was also measured using a Lab Master decagon with an electrolytic measurement cell (CM-2, TECNAL), with the temperature controlled to 25°C^[34,35]. Further, the humidity and total solid content of the nanobiocomposite film were determined by infrared drying using a Shimadzu infrared scale, with the intensity of the radiation adjusted to raise the temperature of the sample to 105°C.

The water solubilities of the different formulations were determined using the method proposed by Gontard et al.^[36]. The traction assays were performed using a texture analyzer (model CT310K, Brookfield) with a maximum load of 10 kN. The ASTM D-882 standard that was adopted for the assays stipulates a loading speed of 0.50 mm/s, temperature of 25°C, trigger load of 7 g, test probe tip of TA3/100, and the use of a TA/TPB probing device. The traction assays were performed using six proof bodies for each sample. The proof bodies were of length 80 mm and width 25 mm, and their strength was calculated by dividing the maximum applied force by the area of the film (width × thickness). The strain at breakage was calculated by dividing the final length by the projection of the probe tip (50 mm) and multiplying by $100^{[37]}$.

2.6 Statistical analysis

The results of the above investigations were expressed in the form of mean \pm standard deviation (n = 3). Statistica[®] 6.0 (StatSoft, Tulsa) was used for the statistical analyses of the results. Analysis of variance (ANOVA) and Tukey's tests were used to determine the significant differences between means (p < 0.05).

3. Results and Discussions

3.1 Characterization of coconut fibers and nanocrystals

Vegetable fibers are mainly composed of cellulose, hemicellulose, and lignin, together with other smaller constituents. The physical and chemical properties of a particular fiber depend on its source and processing. The analysis of the chemical composition of the green coconut fiber in this study revealed cellulose, lignin, and hemicellulose contents of 32.0%, 38.0%, and 0.25%, respectively (Table 2). Trugilho et al.^[38] reported that the lignin and cellulose contents of young plants are variable, and only stabilize with maturation. Corradini et al.^[39] particularly found that the lignin and cellulose contents of a young green coconut fiber varied within 37.2%-43.9% and 31.5%-37.4%, respectively. Similar results were obtained by Rosa et al.^[26,40], who highlighted the high lignin content of green coconut fiber (Table 2) and the potential for its use in the reinforcement of polymeric materials^[39].

Figure 1 shows SEM images of green coconut fiber (Figure 1A) and the extracted cellulose (Figure 1B). The micrographs of the *in natura* fiber (Figure 1A) reveal rugged surfaces covered by layers of wax and extracts, as well as amorphous constituents such as lignin and hemicellulose. The micrographs of the whitened cellulose (Figure 1B) reveal the presence of cellulose fibrils and provide evidence of disaggregated ruggedness, which was due to the conversion of cellulosic elements to fibrils by the removal of poliosis and lignin during the whitening process. Pores and/or orifices can also be observed on the rugged surface. As also evidenced by Alemdar and Sain^[18], these images suggest that the chemical treatment partially removed impurities such as hemicellulose and lignin, which are the "compacting" components of the fibers. The efficient removal of lignin by the whitening of the fibers is extremely important to obtaining nanocrystals by hydrolysis.

The visualization of the suspension of the cellulose nanocrystals obtained from the green coconut fiber (concentration of 0.06 g/10 mL) through polarizers revealed a nematic phase, which was directly produced by the light birefringence and is considered to confirm the presence of nanocrystals (Figure 2). Mesquita et al.[41] and Alves et al.[17] similarly used crossed polarizers to visualize the birefringence phenomenon in a 4% suspension of cellulose nanocrystals obtained from eucalyptus. Birefringence (or optical anisotropy) is a property of transparent crystals that enables them to decompose a light ray into two crossed polarized rays. Observation of the birefringence in a cellulose nanocrystal suspension is thus an effective means of evaluating the nanocrystal formation^[42], and hence assessing the efficiency of the hydrolysis process. It has been observed that suspended cellulose nanocrystals exhibit a tendency to align themselves, probably due to their high solidity and length-to-diameter ratio^[30]. It is also very important to determine whether the cellulose nanocrystals are well dispersed in suspension, because this is required to achieve good results when they are incorporated in a polymeric matrix for mechanical reinforcement^[43,44].

Figure 3 shows the TEM micrographs of the cellulose nanocrystals, from which the length (L), diameter (D),

Table 2. Chemical	composition of gree	n coconut fiber
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Cellulose (%)	Lignin (%)	Reference
32.0	38.0	This study
32.0	40.0	Rosa et al.[40]
37.0	32.5	Rosa et al.[26]
31.5%-37.4%	37.2%-43.9%	Corradini et al.[39]



Figure 1. SEM images of the (A) in natura green coconut fiber and (B) extracted cellulose (B) (Ranges of 1 mm and 100 µm).



Figure 2. Illustration of the suspension of cellulose nanocrystals obtained from green coconut fiber: (A) positions of the Polaroid and sample; (B) sample agitation by birefringence evidencing the presence of the nanocrystals; and (C) magnification of the image in (B) (the arrows indicate the locations of the birefringence phenomena).



Figure 3. TEM images of the nanocrystals obtained from green coconut fiber cellulose, showing (A) agglomeration of the nanocrystals and (B) some isolates (Range: 200 nm).

and aspect ratio (L/D) of the nanocrystals and their state of aggregation were determined. The images confirm the effectiveness of the acid hydrolysis used to produce the cellulose nanocrystals from the green coconut fiber. The images indicate that the aqueous suspensions contained needle-like nanocrystals, consisting primarily of individual fibrils and their aggregates. The cellulose nanocrystals were formed by nanoparticles containing several hydroxyl groups on their surfaces and with a large specific area. Aggregation of the nanocrystals was thus common, as expected, mainly due to the strong hydrogen links formed between cellulose nanoparticles^[45].

The coconut cellulose nanocrystals examined in this study had length L values of 89-320 nm (average of 264.9±23.0 nm) and average diameter D of 8.10±1.21 nm. These results agree well with previously obtained data for cellulose nanocrystals extracted from coconut fibers, namely, L values of 80–500 nm and D values of 4-9 nm^[5,26,40]. Silva et al.^[45] also reported an L value of 145 nm and average D of 1.5 nm for nanocrystals obtained from eucalyptus, while Costa et al.^[6] reported L = 157 nm and average D = 5.7 nm for nanocrystals obtained from licuri fiber. The average aspect ratio (L/D) determined in the present study was 32.7±5.1, which confirms the potential of the coconut cellulose nanocrystals for use as reinforcement agents for polymeric matrixes in the development of composites, as proposed by Rosa et al.^[40] and Machado et al.^[5], who reported L/D values of 39±16 and 38.9±4.7, respectively.

3.2 Characterization of nanobiocomposites

The films produced from the six formulations of nanobiocomposites prepared in this study using different starch types (manioc and potato), as well as the control films, were examined to determine their barrier properties (thickness, water activity, water solubility, and humidity) and mechanical properties (strength and strain at breakage). An attempt was also made to evaluate how these parameters were affected by the addition of the coconut fiber cellulose nanocrystals. The produced starch films were found to be homogeneous, transparent, and visually attractive. Figure 4 shows an example of the manioc starch film with 0.5% nanocrystals (FM05). The determined barrier and mechanical properties of the different films are given in Tables 3 and 4.

Control of the thickness is extremely important to maintaining uniformity of the films and comparing their properties^[46]. Significant difference was observed between the average thicknesses of the FM10 (0.08 mm) and FB (0.12 mm)

control films, as well as among their respective samples. The potato starch films were also found to be thicker than the manioc starch films (see Table 3). The mechanism of the film formation is known to be dependent on the concentration of the solids in the formulation and the amylose content. The potato starch formulation appeared to be more viscous than that of the manioc starch, suggesting a higher amylose concentration of the former, and hence the thicker films. With regard to the formulations containing the same matrix, namely, the same starch content and plasticizer concentration, variation of the drying method (casting) directly resulted in varying film thicknesses, as previously reported^[6,47-49]. Fakhouri et al.^[37] obtained varying film thicknesses between 0.039 and 0.081 mm for starch films plasticized with gelatin, while Reis et al.^[14] obtained thicknesses of 0.11-0.12 mm for manioc starch films containing mango pulp and mate extract.

The water solubility of a film is an important property that determines its applicability for food packaging^[50]. In some cases, total water solubility could be beneficial, such as in the case of semifinished food products that require further cooking. However, when the food item is liquid or aqueous, highly soluble polymeric films are undesirable^[51]. The percentage solubility of the nanocomposites developed



Figure 4. Nanobiocomposites obtained by the incorporation of 0.5% coconut cellulose nanocrystals in a manioc starch matrix plasticized with glycerol (FM05).

Table 3. E	Barrier properties	(average ± standard	deviation) of the	films produced using	different nanobiocomposit	te formulations.
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Formulation	EP	aw	UM	ST	SL
Control FM	0.08±0.01 ^a	0.65±0.01 ^a	15.31±0.78ª	84.68±0.78ª	30.55±0.83ª
FM05	$0.08{\pm}0.01^{a}$	0.58±0.01 ^b	16.21±0.97 ^{b.c}	$83.79 \pm 0.97^{b.c}$	30.54±0.83ª
FM10	0.08±0.01 ^b	0.68±0.01°	18.33±1.29 ^b	81.67±1.29 ^b	31.56±0.80 ^b
FM15	0.09±0.02°	0.71±0.01°	17.98±0.26 ^b	82.01±0.26 ^b	32.16±0.65°
Control FB	$0.12{\pm}0.02^{d}$	0.69±0.01°	16.60± 0.50°	83.39±0.50°	28.30±0.78 ^d
FB05	0.09±0.01°	0.66±0.02°	18.87±0.61b	81.12±0.61b	29.52±0.46 ^d
FB10	0.10±0.02 ^e	0.70±0.01°	18.26±0.91 ^b	81.74±0.91 ^b	25.30±0.36e
FB15	0.10±0.01°	0.68±0.02°	18.46±0.38 ^b	81.53±0.38 ^b	29.17±0.65 ^d

EP - thickness (mm), aw - water activity (%), UM - humidity (%), ST - total solids (%), SL -solubility (%). No significant difference between values with the same superscript letter in a column (p > 0.05), according to Tukey's test with 95% confidence.

Formulation	ST	↑ST	3	$3\downarrow$
Control FM	3.23±0.51ª		26.39±5.04b	
FM05	3.65±0.68 ^b	13.00	23.90±1.67°	9.43
FM10	4.41±0.30°	36.53	$17.54{\pm}0.80^{d}$	33.53
FM15	3.80±0.94 ^b	17.64	24.71±3.18 ^{bc}	6.36
Control FB	4.07 ± 0.20^{d}		30.81±4.44ª	
FB05	8.20±0.52e	101.47	23.71±3.54°	23.04
FB10	5.43 ± 0.24^{f}	33.41	26.50±3.75 ^b	13.98
FB15	4.09 ± 0.64^{d}	0.31	30.40±1.77 ^a	1.33

Table 4. Mechanical properties (average ± standard deviation) of the films produced using different nanobiocomposite formulations.

ST - strength (MPa), \uparrow ST - increase in strength relative to control (%), ε - strain at breakage (%), $\downarrow\varepsilon$ - decrease in strain at breakage relative to control (%). No significant difference between values with the same superscript letter in a column (p > 0.05), according to Tukey's test with 95% confidence.

in the present study varied between 25.30 (FB10) and 32.16 (FM15), and may thus be considered as low-solubility films^[52]. This indicates that they can be used to package a wide variety of food items, provided that their overall performance satisfies relevant legislation.

It was not possible to determine a direct effect of the nanocrystals on the water solubility of the present films. However, the films containing the nanocrystals generally exhibited higher water solubility compared to the control films, although a lower solubility was observed for the film produced from the potato starch matrix containing 10% nanocrystals. The present solubility results are similar to those obtained by Jiang et al.^[50] for films produced from pea starch reinforced with potato starch nanoparticles. Considering the observations of Rubentheren et al.[53] and Pagno et al.[54] all the films containing cellulose nanocrystals were expected to exhibit significantly lower solubility. This is based on the stronger resistance of the films compared to the control films (Table 4), mainly due to the interaction between the nanocrystals and the starch chains. However, only the film produced using formulation FB10 exhibited such decreased solubility.

With regard to the water activity (aw) of the films, the values were found to vary between 0.58 (FM05) and 0.71 (FM15), with significant differences observed for only the films produced from the manioc starch formulations. The aw was not found to be directly affected by the nanocrystal content of the films. The property is, however, an important consideration for food packaging^[55] because the water content affects the growth of microorganisms. The property is thus directly related to the quality conservation of the packaged product. Silva et al.[55] found that starch films containing cellulose nanocrystals had aw values of 0.46-0.63, which substantiates the present observation that formulations containing higher percentages of cellulose nanocrystals produce films with lower aw values. Similar results were obtained by Costa et al.[6], namely, aw values of 0.44-0.49 for manioc films containing propolis extracts and reinforced with licuri cellulose nanocrystals.

All the films of the present study containing coconut cellulose nanocrystals were found to exhibit significantly higher humidity contents compared to the control films. The humidity of the films produced from the control formulation FM was 15.31%, while those of the films

produced from the manioc starch formulations containing cellulose nanocrystals varied between 16.21% and 18.33%. Similar humidity values were obtained for the films produced from the potato starch formulations containing cellulose nanocrystals. The humidity of the films produced from the control formulation FB was 16.60%, while those of the films produced from the other formulations varied between 18.26 and 18.87%. These results indicate that the addition of coconut cellulose nanocrystals dispersed in water increased the humidity of the films. Considering the similar preparation conditions (amount of polymeric matrix, amount of plasticizer, and duration of drying), the higher humidity of the films containing cellulose nanocrystals may be due to the storage for ten days in a desiccator containing saturated sodium chloride solution before the characterization. The films containing the cellulose nanocrystals might have absorbed more water in the desiccator than the control films. Another possible explanation is that the nanocrystals that were incorporated into the plasticized polymeric matrix were dispersed in water, which could have increased the humidity of the films compared to the control films.

Expectedly, the strength and strain at breakage of the starch films were increased by the addition of the cellulose nanocrystals, as indicated in Table 4 and also previously reported^[6,56,57]. The implied enhanced resistance to rupture is desirable and broadens the potential applications of the films to packaging and coating ^[17]. The strengths of the films ranged between 3.23 MPa (control FM) and 8.20 MPa (FB05). An increase of the cellulose nanocrystal concentration was not observed to increase the rupture resistance of the films, indicating that the addition of a large amount of nanoparticles could be counterproductive through their aggregation^[58]. For the manioc starch films, the optimal nanocrystal concentration for mechanical reinforcement was determined to be 1.0%, while it was 0.5% for the potato starch films. In the case of the control films, those produced from potato starch were stronger. Overall, the potato starch films containing cellulose nanocrystals were generally the strongest. The mechanical properties of starch are known to be significantly affected by their botanical origin, specifically the native amylose and amylopectin contents^[59]. The present observed film strengths could thus be explained by the fact that the molecular interactions among the starch chains, plasticizer, and dispersed nanocrystals in the potato starch films were more efficient. This may be due to the better

dispersion of the nanocrystals within the matrix and stronger interfacial adhesion among the components of the complex system, which included the starch, glycerol, and nanocrystals. Similar strength results were obtained by Alves et al.[17], who investigated the incorporation of eucalyptus nanocrystals in maze starch matrixes containing gelatin. They observed better dispersion of the nanocrystals in films with higher gelatin contents, resulting in better mechanical properties. Oun et al.^[60] also observed higher strength values of 22.7-33.7 MPa for agar-based nanocomposites and cotton cellulose nanocomposites. They found that the film traction increased with increasing nanocrystal concentration from 1% to 5%, but decreased significantly (p < 0.05) for 10% nanocrystals. Similar reports have been made by other researchers regarding the tensile properties of films with different matrixes containing cellulose nanocrystals^[61-66].

The determined values of the strain at breakage ranged between 17.54% (FM10) and 30.81% (control FB). It was observed that the two strongest film types (FB05 and FM10) also exhibited the lowest breakage strains. This may be attributed to the higher rigidity of the films due to the strong interactions among the nanocrystals, plasticizer, and polymeric matrix. A small strain at breakage has actually been reported to be an indication of good interaction between the starch and nanocrystals^[54,67,68].

4. Conclusions

The combination of the availability of lignocellulosic fibers and the need for renewable resources for the production of polymers affords strong opportunities for adding value to agricultural products through technological innovation. The results of this study, as well as those of previous works, specifically confirm the viability of developing biodegradable polymeric packaging with adequate characteristics. The nanobiocomposite biodegradable films developed in the present study exhibited desirable barrier and mechanical properties including good transparency, manageability, homogeneity, adequate solubility, and high strength. The addition of cellulose nanocrystals obtained from green coconut fiber to the polymeric matrixes particularly afforded efficient mechanical reinforcement. A low or moderate nanocrystal concentration was found to be more effective depending on the type of polymeric matrix (potato or manioc starch), with higher concentrations inducing undesirable agglomeration of the nanocrystals. The best barrier and mechanic properties were exhibited by the films produced using potato starch, attributable to the better interactions among the polymeric matrix, the cellulose nanocrystals, and the employed plasticizer. It should be noted, though, that the optimal composition of the formulation used to develop the film would depend on the intended use and application technique.

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